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2D Materials



PAPER

Humidity sensing using vertically oriented arrays of ReS₂ nanosheets deposited on an interdigitated gold electrode

RECEIVED
2 August 2016

ACCEPTED FOR PUBLICATION
30 September 2016

PUBLISHED
17 October 2016

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Keywords: humidity sensing, sensitivity, hysteresis, ReS₂ nanosheets, chemical vapor deposition

Abstract

We report a novel humidity sensor featuring vertically oriented arrays of ReS₂ nanosheets grown on an interdigitated gold electrode by chemical vapor deposition. The vertical orientation of the nanosheets is important since it maximizes the exposed surface area for water adsorption/desorption. We find that the resistance of the ReS₂ film decreases sensitively with increasing relative humidity, which we attribute to charge transfer from the absorbed H₂O molecules to the n-doped ReS₂ nanosheets. In addition to high sensitivity, the ReS₂ sensors exhibit fast response/recovery time and excellent reversibility with minimal hysteresis. Moreover, our fabrication approach involving the direct (1-step) growth of the ReS₂ films on an interdigitated electrode (without any transfer using wet chemistry or lithography) greatly simplifies the device architecture and has important practical benefits for the low-cost and scalable deployment of such sensor devices.

Introduction

Recent years have witnessed increasing interest in 2D nanomaterials, such as graphene and transition metal dichalcogenides (TMDs), due to their unique blend of physical, chemical, electronic and optical properties [1, 2]. In particular, 2D materials hold great potential for gas sensing due to their inherently high surface-to-volume ratio. For example, physisorbed small molecules were found to modify carrier density and thus change the conductivity of 2D nanomaterials [3, 4]. While initial efforts were focused on graphene, such graphene based sensors can be problematic in terms of reversibility and selectivity [5, 6]. Of late, TMDs have attracted interest for chemical sensing and revealed better performance in some cases [7]. Very recently, a new member of the TMD family, ReS₂, has attracted significant research interest due to its in-plane anisotropic structure. Unlike the common hexagonal layered dichalcogenides such as MoS₂, WS₂, MoSe₂, and WSe₂, ReS₂ forms a stable distorted 1 T structure

with triclinic symmetry due to an additional valence electron in rhenium [8]. It has been recently reported that the photo-response of ReS₂ transistors could be modulated by adsorbed gas species [9], which suggests that ReS₂ could in-fact serve as a promising 2D material for chemical sensing applications.

In this work, we explore the use of ReS₂ nanosheets for humidity sensing. Detection of humidity is important in various industrial processes and environments [10], such as manufacturing processes control and intelligent control of living environments. Recent efforts have significantly improved the performance of humidity sensors based on 2D materials in terms of their response/recovery times [11–14]. However, for commercial viability of such 2D materials based sensors, it is important to simplify the sensor fabrication process while also minimizing sensor hysteresis. Here, we report the fabrication of a new device architecture featuring vertically oriented arrays of ReS₂ nanosheets grown on an interdigitated gold electrode by chemical vapor deposition (CVD) in a single (1-step) process.

The as-grown ReS_2 films were systematically characterized by scanning electron microscopy (SEM), x-ray diffraction (XRD) and scanning transmission electron microscopy (STEM). Individual ReS_2 sheets were also extracted from the CVD grown films and used as the active element in a field effect transistor (FET) device. The FET tests revealed that the ReS_2 nanosheets exhibit n-type semiconductor characteristics. Therefore when exposed to water, electron injection into the sheets from adsorbed water molecules increases the doping in the material causing a large increase in its electrical conductivity. This increase in the conductivity for the CVD grown ReS_2 films could be detected using simple electrical readout (from the interdigitated electrode device) and correlated to the relative humidity. The ReS_2 sensor reported here showed high sensitivity for humidity detection along with fast response/recovery time and significantly low hysteresis. This can be attributed to the vertical orientation of the ReS_2 sheets, which exposes the full surface area of the 2D material for the rapid adsorption/desorption of water.

We would like to point out that direct growth of the ReS_2 films on an interdigitated electrode (without any transfer) as in our device has important implications for low cost and mass producible devices. So far, most electronic sensors made using TMDs are based on manual mechanical exfoliation of monolayer to few-layer flakes. Such methods involving manual exfoliation are averse to mass production and reproducibility. CVD methods have also been used to grow TMDs on SiO_2/Si or sapphire substrates, but these methods damage the substrate and require transfer of the TMD sheet using chemical wet processing methods. Further, e-beam lithography to electrically address the flakes is complicated and expensive, thus hindering their practical applications. In our device, we overcome the above limitations by direct (1-step) growth of vertically oriented ReS_2 nanosheets onto an interdigitated electrode pattern that is used for electrical readout of the sensor device.

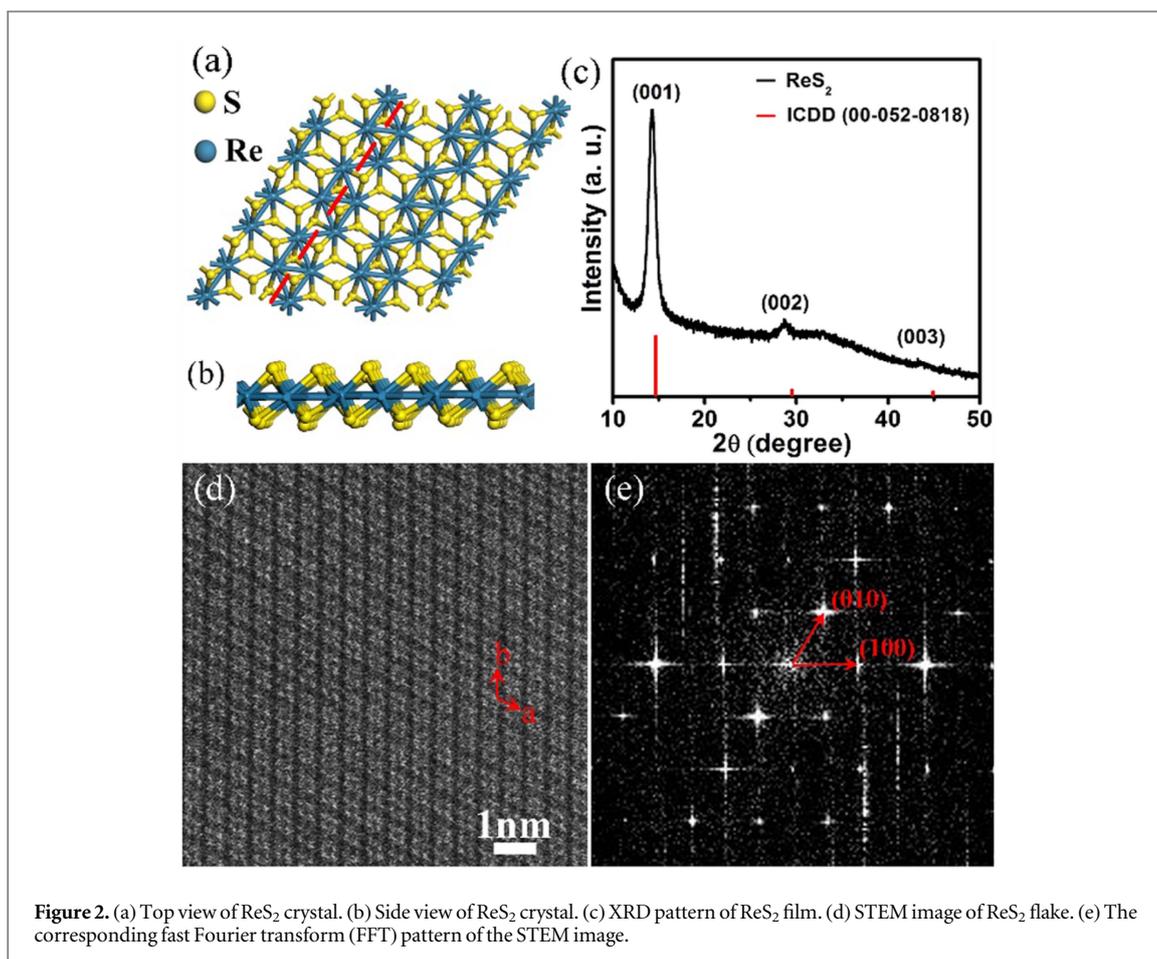
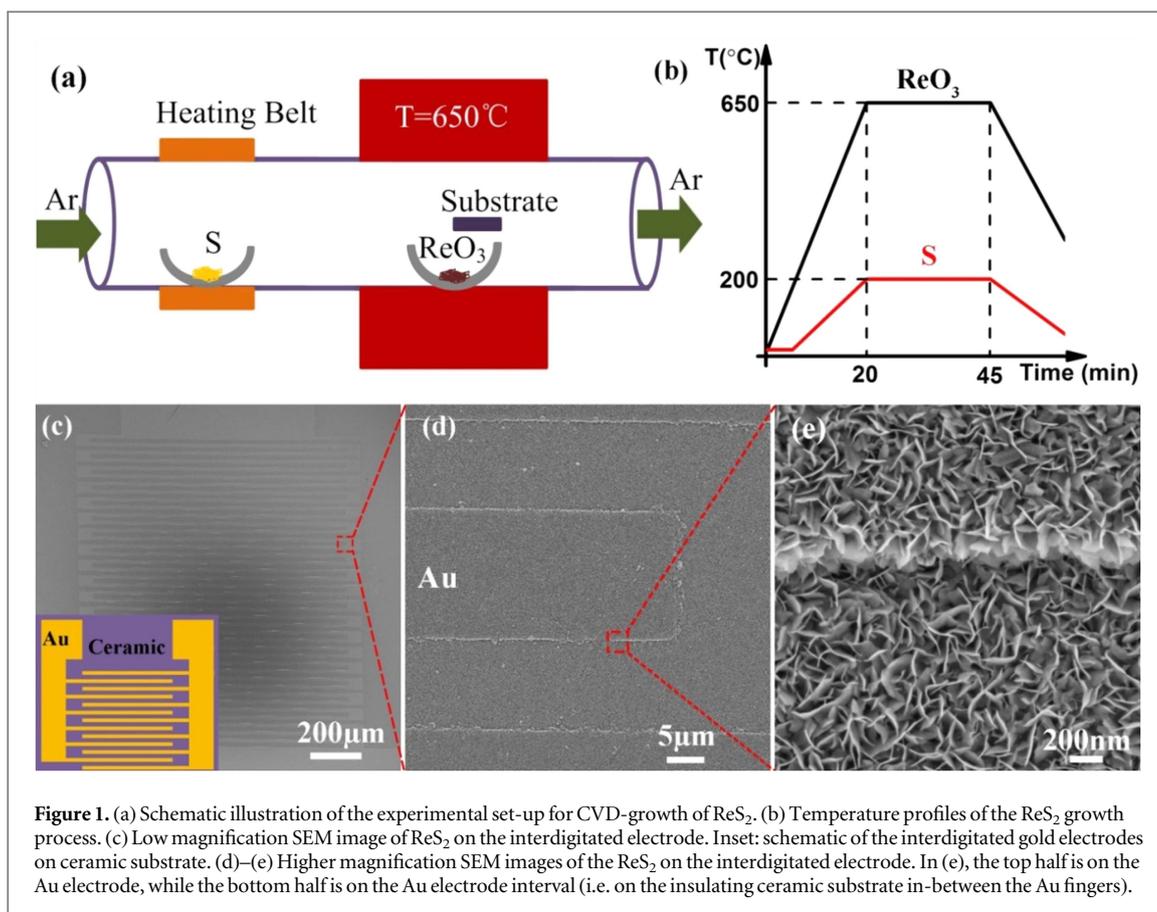
Results and discussions

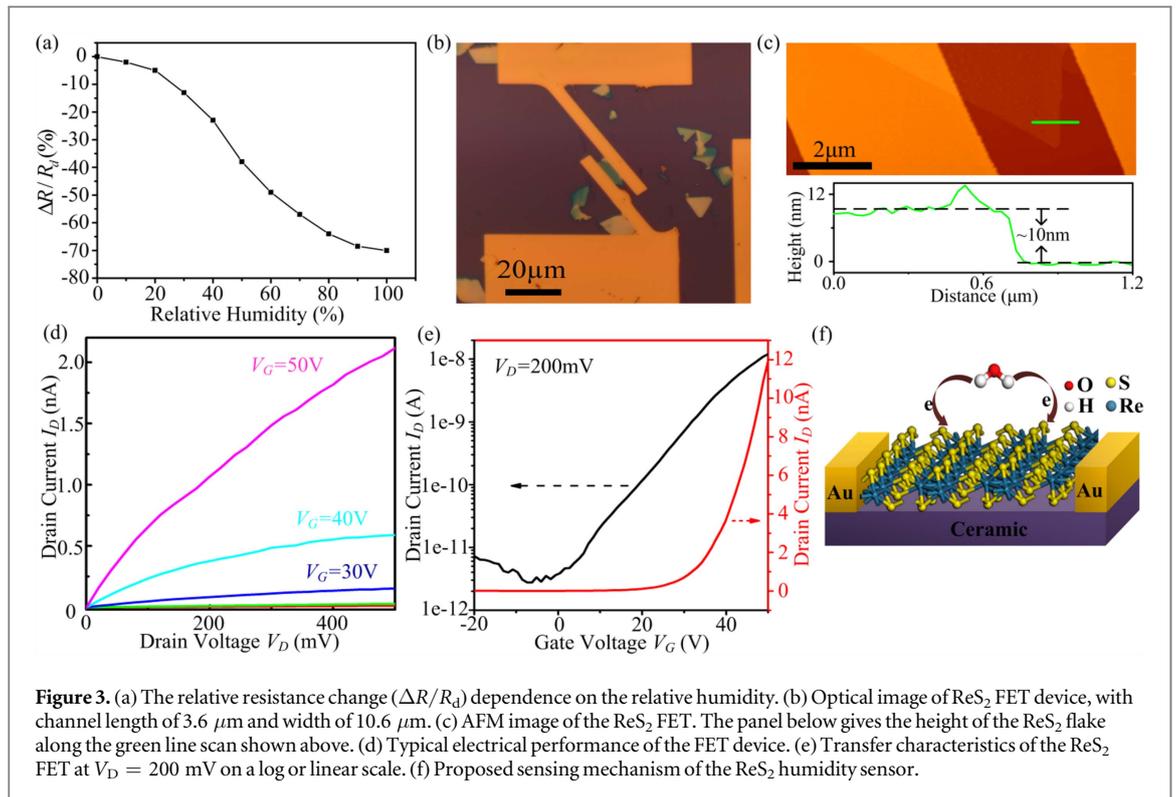
Keyshar *et al* [15], He *et al* [16] and Zhang *et al* [17] have reported the growth of ReS_2 by CVD. In these works, the ReS_2 sheets grew parallel to the growth substrate. By contrast, we have grown vertical ReS_2 by selecting ReO_3 as the precursor for CVD growth and by appropriate tuning of the growth conditions. Figure 1(a) illustrates the scheme for the growth of ReS_2 by CVD method. Briefly, ~ 1 mg rhenium trioxide (ReO_3) powder is placed in an alumina boat with the substrate facing down and the boat is loaded in the center of the furnace. Sulphur powder is placed upstream within the low-temperature zone and heated by a heating belt. During the ReS_2 growth process, the temperatures are controlled according to the profiles

in figure 1(b). Detailed procedure for the growth of ReS_2 can be found in the experimental section. In this work, we have chosen an interdigitated gold electrode pre-patterned on an insulating ceramic wafer as the growth substrate. The schematic of the interdigitated gold electrode is shown in the inset of figure 1(c) with the finger gap of $\sim 10 \mu\text{m}$. To fill up the gap, we have tuned the growth conditions to grow high spatial density, vertically-oriented ReS_2 . Figures 1(c)–(e) are the SEM images of CVD-grown ReS_2 on the interdigitated electrodes. The ReS_2 sheets tend to grow vertically on both Au and ceramic substrates with high packing density, as indicated by the high-magnification SEM image in figure 1(e). Due to high spatial density of ReS_2 sheets, we expected that the adjacent fingers of the interdigitated electrode are electrically connected by the network of ReS_2 nanosheets. The reason for the vertical growth mode that we observe is not clear at this point but could be related to the rapid supply of Re (from fast evaporation of ReO_3 at our CVD growth conditions) that could suppress the atomic migration of Re and S precursors on the surface of the growth substrate [18]. Under such conditions the Re, S precursors could diffuse rapidly on the surfaces of the ReS_2 sheets and attach to the exposed sheets edges contributing to the vertical growth mode.

Figures 2(a) and (b) illustrate the top and side views of the crystal structure of ReS_2 with triclinic symmetry. In striking contrast to most hexagonal TMDs, ReS_2 crystallizes in a distorted 1T structure with Re atoms forming zigzag chains along the b lattice vector direction (indicated by the red dashed line in figure 2(a)). We have carried out XRD theta/2theta scans on the ReS_2 film, as shown in figure 2(c). Peaks at 14.31° , 28.75° , and 43.78° are observed in the XRD pattern, which are in accordance with the (001), (002), (003) peaks of anorthic ReS_2 . The 2theta of these peaks are slightly smaller than the bulk values, which indicates that the CVD crystal in the out of plane direction is relaxed as compared with the bulk. The presence of {001} planes in the theta/2theta XRD pattern can be explained by the bending and tilting of the ReS_2 sheets as shown in figure 1(e). The crystal structure was further studied by using aberration-corrected STEM, with the electron beam perpendicular to the flake basal plane. The high-angle annular dark-field STEM image shown in figure 2(d) matches well with the top view of the ReS_2 structure in figure 2(a), and clearly shows a 001 lattice arrangement for the few-layer ReS_2 flake. The corresponding fast Fourier transform image (figure 2(e)) displays a quasi-hexagonal pattern formed by (100), (010) reflections, which is in good accordance with previous studies [19].

The humidity sensing setup is shown in figure 5 (experiment section). The as-prepared humidity sensor (figure 1) without any modification was placed in an environmental chamber. The humidity level in the environmental chamber was controlled by adjusting the mixing ratio of dry to wet nitrogen obtained by



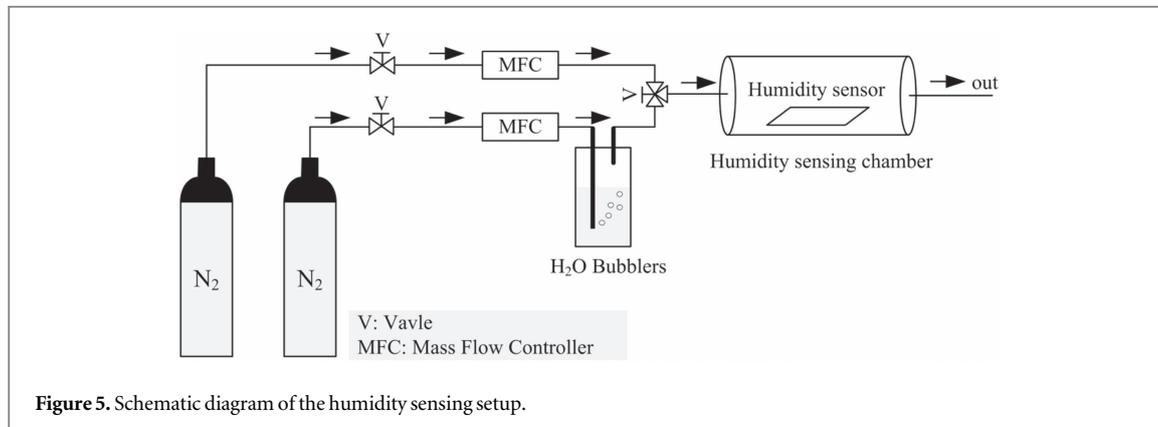
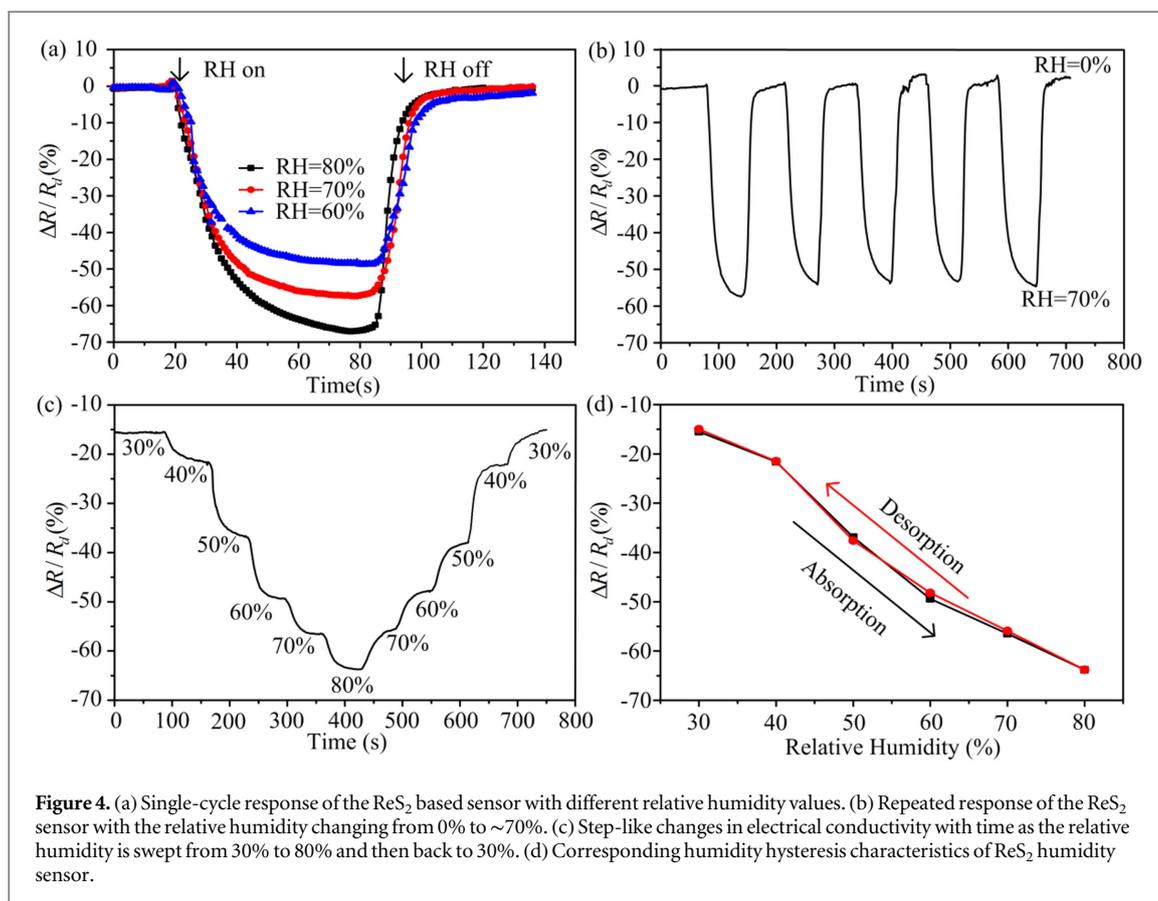


bubbling a N₂ stream through a flask with water. Details of the humidity sensing method are provided in the experimental section. Figure 3(a) illustrates the relative resistance change $\Delta R/R_d$ dependence on the relative humidity (RH), where $\Delta R = R - R_d$ and R , R_d is the resistance of the ReS₂ based sensor in humidity gas and dry nitrogen, respectively. It can be seen that the resistance decreases monotonically with increasing relative humidity. In order to investigate the underlying mechanism that is responsible for this resistance change, we extracted an individual sheet from the ReS₂ film (see experimental section for details) and constructed a back-gated FET device on a conventional SiO₂/Si substrate.

The optical image of the ReS₂ FET device is shown in figure 3(b) while the atomic force microscopy (AFM) image of the ReS₂ FET is presented in figure 3(c). The AFM line scan shows that the thickness of the ReS₂ channel is ~ 10 nm, which indicates that the ReS₂ flake is about 14 layers [8, 9, 15, 16]. The typical electrical performance of the FET device is presented in figure 3(d). It is found that the drain current increases with gate voltage, which indicates that the fabricated FET displays n-type semiconducting behavior consistent with previous results for CVD-grown and mechanically exfoliated ReS₂ [9, 15, 16, 19–22]. From the transfer curve shown in figure 3(e), the threshold voltage is extrapolated as ~ 33 V from the linear scale curve, and the on/off ratio is $\sim 10^4$ in the testing gate voltage range. The field effect carrier mobility can be calculated by using the equation: $\mu = \left[\frac{dI_D}{dV_G} \right] \times \left[\frac{L}{WC_i V_D} \right]$, where I_D is the drain current,

V_G is back-gate voltage, V_D is the drain voltage of 100 mV, channel length $L = 3.6$ μm , channel width $W = 10.6$ μm and the capacitance C_i is estimated to be ~ 11.6 nF cm⁻² ($C_i = \epsilon_0 \epsilon_r / d$, where ϵ_0 is the dielectric constant, $\epsilon_r = 3.9$ and $d = 300$ nm). The calculated field effect mobility of ReS₂ at $V_G = 50$ V is ~ 0.32 cm² V⁻¹ s⁻¹. Since our ReS₂ film exhibits n-type behavior, the humidity sensing mechanism can be explained by n-doping of the ReS₂ sheets as shown schematically in figure 3(f). The electrical conduction of the ReS₂ film is dominated by electrons. H₂O molecules absorbed on the surfaces of the vertically oriented ReS₂ sheets will donate electrons to the n-type ReS₂, which increases the carrier concentration (electrons) in the channel between the interdigitated gold electrodes, thereby increasing the conductance. Note that this is in contrast to the carbon nanotube (CNT) based humidity sensor in Han *et al's* work [23], where the resistance increased with relative humidity due to the reduction of hole density in the p-type CNT.

Figure 4(a) illustrates a typical single-cycle response of the ReS₂ based sensor with different relative humidity values. It is observed that the resistance of the ReS₂ sensor decreases rapidly on exposure to moisture. After the humidity gas is switched off, the resistance quickly recovers to almost 100% of the original value. The response and recovery behavior is an important metric to evaluate the performance of a humidity sensor [24]. The response and recovery time of a sensor are defined as the time taken for the resistance change to reach $\sim 90\%$ of the steady state value. For our moisture sensor, the response and recovery time (figure 4(a)) are ~ 20 s and ~ 10 s, respectively. As



compared to the previously reported CNT film [23], ZnO nanosheets [24] and VS₂ nanosheets [25] based humidity sensors, the ReS₂ sensor in this work exhibits faster response and recovery behavior. The faster response is presumably a result of the vertical orientation of the ReS₂ in our film, which exposes the entire surface of the nanosheets enabling rapid adsorption and desorption of water. Although some previous humidity sensors [11–14] have shown better performance for response/recovery time, our sensor is produced by a relatively simple (1-step) manufacturing process and exhibits excellent reversibility (see figures 4(c) and (d)). Figure 4(b) reveals reversible switching behavior for the ReS₂ sensor when switching

on and off the humidity (RH = 70%). In every cycle, the sensor responds quickly (in ~20 s) upon exposure to humidity and recovers (in ~10 s) close to the original point after the humidity is removed. We have verified this performance for several different sensor devices indicating that the results are reproducible.

Hysteresis is a common limitation for commercial metal oxide based humidity sensors [24, 26], and results from incomplete water desorption. It is crucial to minimize the hysteresis effect for practical applications. Hysteresis can be quantified by using the relation: $|\frac{\Delta R_{\text{abs}}}{R_d} - \frac{\Delta R_{\text{des}}}{R_d}| / (\frac{\Delta R_{\text{abs}}}{R_d}) \times 100\%$, where $\Delta R_{\text{abs}}/R_d$ and $\Delta R_{\text{des}}/R_d$ are the relative resistance change in the absorption and desorption process, respectively. In

order to study the hysteresis effect of our ReS₂ sensor, the relative humidity was swept from 30% to 80% and back to 30% in steps of 10%. Figure 4(c) presents the step-like changes with time in the response as the relative humidity is varied, and figure 4(d) shows the corresponding hysteresis characteristics, showing that the ReS₂ sensor exhibits excellent reversibility (with hysteresis value <1.5%). This indicates that H₂O is rapidly desorbed from the ReS₂ and there is minimal residual water left on the ReS₂ nanosheets. Compared to TiO₂/LiCl [26] and supramolecularly modified graphene composite [13] based humidity sensors, the ReS₂ sensor device shows significantly less hysteresis effect.

Conclusion

In summary, we report a robust and high performance humidity sensor device featuring vertically oriented and densely packed ReS₂ nanosheets grown on an interdigitated electrode by CVD. The fabricated ReS₂ sensors exhibited high sensitivity, fast response and recovery times, and excellent reversibility with little hysteresis effect. The direct (1-step) growth of the ReS₂ films on an interdigitated electrode (without any transfer using wet chemistry or e-beam lithography) as in our device has important practical benefits for mass production and operation of such sensor devices in a reproducible manner.

Experimental section

Preparation of ReS₂ thin film

The substrate is an interdigitated gold electrode (width: ~20 μm, spacing: ~10 μm, NanoSPR Devices) pre-patterned on an insulating ceramic wafer. The substrate was washed thoroughly with acetone followed by methanol and iso propyl alcohol and dried with nitrogen. Then few-layer ReS₂ nanosheets were synthesized in a horizontal tube furnace equipped with a 1 inch diameter quartz tube. An alumina boat containing 200 mg of sulfur (≥99.9% Alfa Aesar) was placed upstream in the quartz tube. The substrate was put face down above another alumina boat containing ~2 mg of ReO₃ (≥99.9% AlfaAesar) and placed in the center of the furnace. The quartz tube was pumped and flushed with nitrogen gas in order to remove air. The furnace was then ramped to ~650 °C in ~11 min and held at ~650 °C for ~25 min with a constant flow rate of argon at ~20 sccm under atmospheric pressure. After the growth cycle was completed, the system was cooled down to room temperature.

FET device fabrication and electronic performance

The ReS₂ flakes are transferred to the 300 nm SiO₂/n++Si substrates prior to device fabrication. A drop of 2-propanol was deposited over the ReS₂ flakes, and the SiO₂/Si substrate was placed directly on the ReS₂ sample, pulling it into contact with the ReS₂ flakes

while naturally drying. Some ReS₂ flakes attached on the SiO₂/Si substrate after separation. The ReS₂ was patterned using Nabity Pattern Generation System e-beam lithography equipped in Carl Zeiss Supra SEM, and Ti/Au (5 nm /50 nm) electrodes were deposited by e-beam evaporation. The n++ Si was used as the back gate. The typical *I*-*V* characteristics were measured by a HP4155B semiconductor parameter analyzer.

Characterization method

SEM images were taken using Carl Zeiss Supra SEM, and the AFM images were taken by PSI XE100 AFM in contact mode. The AFM tip (μmash, HQ: CSC17/AL BS) used has a tip radius of 8 nm, a force constant of 0.18 N m⁻¹, and a resonant frequency of 13 kHz. Aberration-corrected STEM imaging experiments were performed with a Nion UltraSTEM-100, equipped with a cold field emission electron source and a corrector of 3rd and 5th order aberrations [27]. The microscope was operated at 100 kV accelerating voltage and the convergence semi-angle of the incident probe was set to ~30 mrad. The high-angle annular dark field images—also known as Z-contrast—were collected from ~86 to 200 mrad half-angle range.

Humidity sensing method

The humidity sensing setup is shown in figure 5. The as-prepared humidity sensor (figure 1) was placed in an environmental chamber for the tests. The wet gas flow was obtained by bubbling a N₂ stream through a flask with water. The humidity level in the environmental chamber was controlled by adjusting the mixing ratio of dry to wet nitrogen using a mixing system equipped with mass flow controllers. The total flow was set to ~500 sccm in our experiments. The resistance of sensor was measured by an Agilent 34401A instrument with four-probe method. The four-probe method eliminates the effect of contact resistances. All tests were conducted at room temperature and atmospheric pressure.

Acknowledgments

NK and TML acknowledge funding support from the USA National Science Foundation (Award Numbers 1234641, 1435783, 1510828, and 1608171). Microscopy research was conducted as part of a user proposal through ORNL's Center for Nanophase Materials Sciences, which is a US Department of Energy, Office of Science User Facility (JCI). AY and MR acknowledge support from the National Key Basic Research Program of China (973Program) (2015CB251002), National Science Foundation of China (Grant No.51521065 and 51221005), China Postdoctoral Science Foundation (2015M572558) and the State Key Laboratory of Electrical Insulation and Power Equipment (No. EIPE16307).

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